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An ionic Diels–Alder route to *cis*-fused octalins containing an all-carbon quaternary stereocenter in an angular position

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ABSTRACT

Article history: Received 18 December 2009 Accepted 21 December 2009 Available online 4 January 2010 We describe herein the development of efficient and stereoselective synthetic routes to a series of *cis*octalins possessing an all-carbon quaternary center in an angular position.

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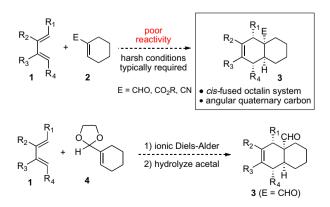
Polycyclic *cis*-fused octalin systems bearing angular quaternary carbon centers, of the type **3**, can be of considerable value in organic synthesis.¹ In principle, this type of functionality could be rapidly reached through Diels–Alder cycloaddition between a substituted butadiene (cf. **1**) and a cycloalkene dienophile, conjugated to an exocyclic activating group (E). However, we and others have observed that cyclic dienophiles of the type **2**^{2.3} generally exhibit poor reactivity in the Diels–Alder cycloaddition, necessitating exposure to harsh reaction conditions (such as high reaction temperature and prolonged reaction time) or the use of a highly reactive diene. In the course of a program underway in our laboratory, focused on the investigation of new applications of Diels–Alder strategies, we sought to gain ready access to generic octalin motifs of the type **3**.

In the light of the difficulties associated with applying the traditional Diels–Alder reaction to the problem at hand, we considered the possibility of adopting the ionic variation en route to **3**. Gassman et al. were the first to describe the ionic Diels–Alder reaction of α , β -unsaturated ketals, as a useful means by which to prepare cycloadducts under mild reaction conditions.⁴ More recently, Grieco et al. developed a modified protocol, which allows α , β -unsaturated ketals to undergo facile ionic Diels–Alder cycloaddition, in the presence of 4.0 M LiClO₄ in Et₂O containing 1.0 mol % camphorsulfonic acid (CSA).⁵ As described below, we sought to extend the Grieco ionic Diels–Alder protocol to encompass acetal substrates of the type **4**.⁶ The development of this capability could well be a useful method for the preparation of highly substituted, *cis*-fused octalin systems of the type **3** (Scheme 1).⁷

In the event, it was found that, upon exposure to 4.0 M LiClO_4 in Et₂O and 1.0 mol % CSA, diene **5** and dienophile **4** underwent the desired ionic Diels–Alder cyclization to afford a ~3:1 mixture of the direct acetal cycloadduct and the aldehyde product (**9**), which presumably arises through hydrolysis of the acetal intermediate. In

situ hydrolysis of the acetal was suppressed through the use of freshly distilled dienes in the Diels-Alder step. Subsequent hydrolysis of the acetal cycloadduct delivered the target aldehyde, 9, in 53% yield over two steps (Table 1, entry 1). This protocol was readily extended to a series of acyclic dienes, as shown in Table 1, and moderate overall yields were obtained (41-58% over two steps). When diene 7, possessing a weakly directing group at the 2-position, was investigated, moderate regioselectivity was observed (4:1, para:meta). By contrast, diene 8, possessing substitution at the 1-position, demonstrated very high regioselectivity, and afforded adduct 12 with 10:1 endo selectivity. It is of note that each of the reactions presented in Table 1 were conducted at ambient temperature and pressure. A control experiment, performed in the absence of CSA, resulted only in the recovery of unreacted starting material, confirming the critical role played by Brønsted acid in the reaction sequence.⁸

In an effort to further explore the utility of this approach to *cis*fused octalin motifs, we sought to further functionalize aldehyde adducts **9** and **10**. As shown in Scheme 2, the aldehydes were readily converted to the corresponding nitriles (**13** and **14**, respec-



Scheme 1. A proposed route to cis-fused octalins bearing an angular substituent.

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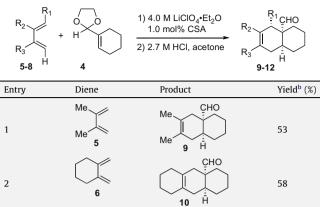
^{0040-4039/\$ -} see front matter \circledcirc 2009 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2009.12.127

Table 1

3

4

Ionic Diels-Alder reactions and hydrolysis of the resulting acetals^a



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е 11 н Ме СНО 56^c

41^d

We also thank Dr. Yasuhiro Itagaki (Mass Spectral Core Facility, Columbia University) for mass spectral analysis. **Supplementary data**

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.tetlet.2009.12.127.

tively), in a two-step protocol, via the intermediate formation of an

aldoxime function.⁹ In another demonstration of the synthetic versatility of these cycloadducts, aldehydes **9** and **10** were advanced

to the octalin esters, **15** and **16**, through Pinnick oxidation¹⁰ fol-

method which is superior to those currently in use for the preparation of a series of *cis*-fused octalin derivatives containing an all-car-

In summary, we have described herein the development of a

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lowed by esterification of the resulting carboxylic acids.

bon guaternary stereocenter at the ring-junction carbon.

References and notes

Acknowledgments

- For recent reviews, see: (a) Trost, B. M.; Jiang, C. Synthesis 2006, 369–396; (b) Douglas, C.; Overman, L. E. Proc. Natl. Acad. Sci. U.S.A. 2004, 101, 5363–5367; (c) Corey, E. J.; Guzman-Perez, A. Angew. Chem., Int. Ed. 1998, 37, 388–401.
- (a) Szmuszkowicz, J.; Bergmann, E. D. Bull. Res. Counc. Israel 1953, 3, 93–95; (b) Bergmann, E. D.; Becker, A. J. Am. Chem. Soc. 1959, 81, 221–225.
- 3. Danishefsky, S.; Kitahara, T. J. Org. Chem. 1975, 40, 538-539.
- (a) Gassman, P. G.; Singleton, D.; Wilwerding, J. J.; Chavan, S. J. Am. Chem. Soc. 1987, 109, 2182–2184; (b) Gassman, P. G.; Chavan, S. P. J. Org. Chem. 1988, 53, 2392–2394; (c) Gassman, P. G.; Chavan, S. P. Tetrahedron Lett. 1988, 29, 3407– 3410; (d) Gassman, P. G.; Chavan, S. P. J. Chem. Soc., Chem. Commun. 1989, 837– 839.
- (a) Grieco, P. A.; Collins, J. L.; Handy, S. T. Synlett **1995**, 1155–1157; For the application of Grieco's condition to polycyclic systems, see: (b) Hailes, H. C.; Isaac, B.; Javaid, M. H. *Tetrahedron* **2001**, *57*, 10329–10333; (c) Lee, J. H.; Kim, W. H.; Danishefsky, S. J. *Tetrahedron Lett.* **2009**, *50*, 5482–5483.
- (a) Fateeva, A.; Pécaut, J.; Bayle, P.-A.; Maldivi, P.; Dubois, L. Eur. J. Org. Chem. 2009, 3845–3848; (b) Fateeva, A.; Calborean, A.; Pécaut, J.; Maldivi, P.; Marchon, J.-C.; Dubois, L. Tetrahedron 2008, 64, 10874–10881; (c) Eike, H.; Knochel, P. Angew. Chem., Int. Ed. 2001, 40, 3022–3025; (d) Eike, H.; Knochel, P. Org. Lett. 2001, 3, 127–130.
- 7. To the best of our knowledge, the known acetal **4** never been utilized in any Diels–Alder Reactions. For the ionic Diels–Alder reactions of acetals of α , β -unsaturated aldehydes, see: (a) Kumareswaran, R.; Vankar, P. S.; Reddy, V. R.; Pitre, S. V.; Roy, R.; Vankar, Y. D. *Tetrahedron* **1999**, *55*, 1099–1110. and pertinent references cited therein; (b) Sammakia, T.; Berliner, M. A. J. Org. Chem. **1994**, *59*, 6890–6891.
- Diels–Alder reaction of 4 (1.0 equiv) with 2,3-dimethyl-1,3-butadiene (5.0 equiv) under thermal condition (0.5 M in toluene-d₈, 5.0 mol % of 2,6-di*tert*-butyl-4-methylphenol, 150 °C, 40 h) afforded a complete recovery of the acetal 4.
- (a) We note that the isonitrile adducts, **13** and **14**, have served as useful reference compounds in the context of a separate Diels-Alder related investigation. Kim, W. H.; Lee, J. H.; Aussedat, B.; Danishefsky, S. J., manuscript in preparation.; (b) Kim, W. H.; Lee, J. H.; Danishefsky, S. J. J. Am. Chem. Soc. **2009**, *131*, 12576-12578.
- 10. Bal, B. S.; Childers, W. E., Jr.; Pinnick, H. W. Tetrahedron 1981, 37, 2091-2096.

 a Diels–Alder reactions were carried out with 1.0 mmol of 4, 5.0 mmol of diene in 4.0 M LiClO₄ (16.0 mmol) in Et₂O (4.0 mL), in the presence of 0.5 M CSA in THF (20 μ L, 1.0 mol %) at room temperature for 15–17 h.

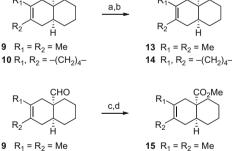
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^b Isolated yields of the cycloadducts over two steps.

СНО

^c p-Directed/m-directed = 4:1.

^d Endo/exo = 10:1.



9 R₁ = R₂ = Me **10** R₁, R₂ = -(CH₂)₄-

Scheme 2. Derivatization of angular aldehydes. Reagents and conditions: (a) H_2 NOH-HCl, NaOAc, THF, reflux, 2 h; (b) SOCl₂, DMF, 0–23 °C, 4 h, 60% over two steps (*cis*-13), 56% over two steps (*cis*-14); (c) NaClO₂, NaH₂PO₄, 2-methyl-2-butene, *t*-BuOH/H₂O (4:1, v/v), 23 °C, 2.5 h; (d) TMSCHN₂, benzene/CH₃OH (3:1, v/v), 0–23 °C, 1.5 h, 99% over two steps (*cis*-15), 88% over two steps (*cis*-16).

16 $R_1, R_2 = -(CH_2)_4$